

## Assignment 6, due November 7

**Corrections:** (none yet)

See notes on the Ito Integral and diffusions posted on the Resources page.

1. Consider the incorrect approximation formula for  $\int_0^t W_s dW_s$ :

$$Y_t^{\Delta t} = \sum_{t_k < t} W_{t_{k+1}} (W_{t_{k+1}} - W_{t_k}) . \quad (1)$$

Repeat (changing as necessary) the steps that lead to

$$X_t^{\Delta t} \rightarrow X_t = \frac{1}{2} (W_t^2 - t)$$

to find a formula for

$$Y_t = \lim_{\Delta t \rightarrow 0} Y_t^{\Delta t} .$$

Show that  $Y_t \neq X_t$ . Explain the difference by calculating

$$E[W_{t_{k+1}} (W_{t_{k+1}} - W_{t_k})] \neq E[W_{t_k} (W_{t_{k+1}} - W_{t_k})] .$$

This calculation predicts a formula for  $Y_t - X_t$ . Find it.

2. The *urn process* (or one of the processes called urn process) has a number of red and blue balls totaling  $n$ . The number of red balls at time  $t_k$  is  $N_k$ . At each time, you remove one of the balls (chosen independently and at random, with each ball equally likely to be chosen) and replace it with a new independent ball with probability  $p$  to be blue and  $1 - p$  to be red.
  - (a) Calculate the probabilities for  $N \rightarrow N + 1$  and  $N \rightarrow N - 1$  (and therefore  $N \rightarrow N$ ) in one step of the process. Use this to calculate  $E[\Delta N | \mathcal{F}_k]$  and  $E[(\Delta N)^2 | \mathcal{F}_k]$ .
  - (b) Write an SDE for  $N_t$  assuming  $\Delta t = 1$  and  $t_k = k\Delta t$ . Take  $dt = 1$  in this calculation. This calculation is sort of correct and sort of incorrect. We will refine it by using systematic *scalings* next week.
3. Consider a chemical system in which there are  $N_t$  copies of a certain kind of molecule at time  $t$ . In a time  $dt$ , there is probability  $\lambda dt$  that a new molecule of this type “arrives” (is produced by some chemical reaction). Molecules of this type are removed by combining with each other to form

*dimers* (compound molecules consisting of two of the originals bound together). The probability that a given molecule can find a partner and form a dimer is  $\mu(N_t - 1)dt$ . Here  $N_t - 1$  is the number of other molecules that this one could combine with. Write an approximate SDE for this process, assuming  $N_t$  is large. The difference between  $N_t$  and  $N_t - 1$  should not matter if  $N$  is large.

4. The *Ornstein Uhlenbeck* process is the one we saw a few weeks ago:  $dV_t = -\gamma V_t dt + \sigma dW_t$ . Suppose  $X_0 = 0$  and  $dX_t = V_t dt$ . Show that for large  $t$ ,  $E[X_t^2] \approx Dt$  and find a formula for  $D$  in terms of  $\gamma$  and  $\sigma$ . Hint: express  $X_t$  as a double integral of  $dW_t$  and reverse the order of integration. Use the previous expression for  $V_t$  as the first step. This formula was first derived by Einstein in his model of Brownian motion.